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Date Mailed : March 31, 2005

Patent No. : 6,755,925 B1
Inventor : Brown, Paul
Patent Issued : January 29, 2004
Title : METHOD OF RESISTING CORROSION IN METAL REINFORCING ELEMENTS...
Docket No. : 287122-00001-2

Re: Request for Certificate of Correction

Consideration has been given your request for the issuance of a certificate of correction for the above-identified patent under the provisions of Rule 1.322.

The inventors' names are printed in accordance with the typed names on the Declaration, therefore the error was the result of applicants' failure to comply with the requirement that the correct names be indicated on the Declaration or Oath. No correction is in order here under the provisions of Rule 1.322.

In view of the foregoing, your request for certificate of correction is hereby denied. However, further consideration will be given these matters, upon receipt of a request for certificate of correction under the provisions of 1.323, accompanied by the appropriate fee which is presently \$100.

A certificate of correction will be issued to correct the error(s) noted in your request.

Future correspondence concerning this matter should be filed and directed to Decisions & Certificates of Correction Branch.

Commissioner of Patents
P.O. Box 1450
Alexandria, VA 22313-1450
Attn: Certificates of Correction Branch

A handwritten signature in cursive script, appearing to read "David Irvine".

David Irvine
Cecilia Newman, Supervisor
Decisions & Certificates of Correction Branch
703/308-9590 or 703/305-4362

Arnold B. Silverman
600 Grant Street, 44th Floor
Pittsburgh, Pa. 15219

CBN/ di

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 6,755,925 D1

DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

NO
Cover Page, 1st Col. "~~Paul Brown~~" should read --Paul W. Brown--.

Col. 3, lines 9, 10, 11 " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$; $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$; and $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ " should read -- $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$; $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$; and $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ --.

Col. 3, line 17 " $3\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}$ " should read -- $3\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}$ --.

Col. 3, line 22, " $\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})n\text{H}_2\text{O}$ " should read -- $\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})n\text{H}_2\text{O}$ --.

Col. 4, line 1, " Fe^{2+} to a Fe^{3+} " should read --" Fe^{2+} to a Fe^{3+} --.

Col. 6, line 12 and 13, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O} + 2\text{Cl} \rightarrow 3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot n\text{H}_2\text{O} + 2\text{NO}_2$ " should read -- $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O} + 2\text{Cl} \rightarrow 3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot n\text{H}_2\text{O} + 2\text{NO}_2$ --.

Col. 6, lines 15 and 21, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ " should read -- $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ --.

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MAILING ADDRESS OF SENDER:

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Pittsburgh, PA 15219

PATENT NO. 6,755,925

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UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 6,755,925 B1

DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

Col. 6, line 38 " $(3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O})$ " should read $-(3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O})-$.

Col. 6, line 41 " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot 10\text{H}_2\text{O}$ " should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot 10\text{H}_2\text{O}-$.

Col. 6, lines 51 and 52, " $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O} + 2\text{Cl}^- \rightarrow 3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot n\text{H}_2\text{O} + 2\text{NO}_2^-$ ", should read $-3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O} + 2\text{Cl}^- \rightarrow 3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{CaCl}_2 \cdot n\text{H}_2\text{O} + 2\text{NO}_2^- -$.

Col. 6, line 54, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ ", should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}-$.

Col. 6, line 67, " $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ ", should read $-3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}-$.

Col. 7, line 1, "24; $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$ wherein $n=0$ to 24; and $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$ wherein $n=0$ to 24.", should read
--24;

$3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$ wherein $n=0$ to 24; and
 $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$ wherein $n=0$ to 24.--

Col. 7, line 3, " $3\text{Me(II)} \cdot \text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me(II)}(\text{anion})_2 \cdot n\text{H}_2\text{O}$ ", should read $-3\text{Me(II)} \cdot \text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me(II)}(\text{anion})_2 \cdot n\text{H}_2\text{O}-$.

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UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 6,755,925 B1

DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

Col. 7, lines 10-11, " $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot \text{nH}_2\text{O}$ or $(3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O})$," should read $-3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot \text{nH}_2\text{O}$ or $(3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O})-$.

Col. 7, line 13, " $\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})$," should read, $-\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})-$.

Col. 7, line 25 " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}$," should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}-$.

Col. 7, line 25-26, " $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}$ " should read $-3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}-$.

Col. 8, line 1, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3$," should read $--3\text{CaO} \cdot \text{Al}_2\text{O}_3--$.

Col. 8, line 1, " $\text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}$," should read $--\text{Ca}(\text{NO}_2)_2 \cdot \text{nH}_2\text{O}-$.

Col. 8, line 10, " $\text{CaO} \cdot \text{Al}_2\text{O}_3$," should read $--\text{CaO} \cdot \text{Al}_2\text{O}_3--$.

Col. 8, line 19, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot \text{nH}_2\text{O}$ " should read $--3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot \text{nH}_2\text{O}-$.

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MAILING ADDRESS OF SENDER:

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UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 6,755,925 B1

DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

Col. 8, line 23, "3CaO.Al₂O₃" should read --3CaO • Al₂O₃--.

Col. 8, line 37, "CaO.Al₂O₃" should read -- CaO • Al₂O₃--.

Col. 8, line 55, "3CaO.Al₂O₃.Ca(NO₃)₂.nH₂O" should read --3CaO • Al₂O₃ • Ca(NO₃)₂ • nH₂O--.

Col. 8, line 59, "3CaO.Fe₂O₃.CaCl₂.nH₂O" should read --3CaO • Fe₂O₃ • CaCl₂ • nH₂O --.

Col. 8, line 60-61 and 64, "3CaO.Fe₂O₃.Ca(NO₃)₂.nH₂O" should read --
3CaO • Fe₂O₃ • Ca(NO₃)₂ • nH₂O--.

Col. 8, line 61 and 65, "3CaO.Fe₂O₃.Ca(NO₃)₂.nH₂O" should read --
3CaO • Fe₂O₃ • Ca(NO₃)₂ • nH₂O--.

Col. 8, line 63, "(3CaO.Al₂O₃.CaCl₂.10H₂O)" should read --(3CaO • Al₂O₃ • CaCl₂ • 10H₂O)--.

Col. 8, line 66, "2CaO.Fe₂O₃" should read --2CaO • Fe₂O₃--.

Col. 9, lines 1 and 4, "2CaO.Fe₂O₃" should read --2CaO • Fe₂O₃--.

Col. 9, line 9, both instances "2CaO.Fe₂O₃" should read --2CaO • Fe₂O₃--.

Col. 9, line 13, "3CaO.Fe₂O₃.Ca(NO₃)₂.nH₂O" should read --3CaO • Fe₂O₃ • Ca(NO₃)₂ • nH₂O--.

Col. 9, line 16, "2CaO.Fe₂O₃" should read --2CaO • Fe₂O₃--.

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PATENT NO. : 6,755,925 B1

DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

Col. 9, line 21 and 22, "3Me(II)O.R₂O₃Me(II)(anion)₂.nH₂O" should read --
3Me(II)O • R₂O₃Me(II)(anion)₂ • nH₂O--.

Col. 9, line 41, "ther" should read --there--.

Col. 10, line 54 and 55, "3CaO.Al₂O₃.Ca(NO₂)₂.nH₂O" should read --
3CaO • Al₂O₃ • Ca(NO₂)₂ • nH₂O--.

Col. 10, line 64 and 65 "3CaO.Al₂O₃.Ca(NO₃)₂.nH₂O" should read --
3CaO • Al₂O₃ • Ca(NO₃)₂ • nH₂O--.

Col. 11, lines 56 and 57,
"2NaAlO₄+3Ca(OH)₂+Ca(NO₂)₂→3CaO.Al₂O₃.Ca(NO₂)₂.nH₂O+2NaOH" should read --
2NaAlO₄+3Ca(OH)₂+Ca(NO₂)₂→3CaO • Al₂O₃ • Ca(NO₂)₂ • nH₂O+2NaOH--.

Col. 11, lines 61 and 62,
"2NaAl₂O₄+4Ca(OH)₂+2NaNO₂→3CaO.Al₂O₃.Ca(NO₂)₂.nH₂O+4NaOH"
should read --2NaAl₂O₄+4Ca(OH)₂+2NaNO₂→3CaO • Al₂O₃ • Ca(NO₂)₂ • nH₂O+4NaOH--.

Col. 12, line 9, "CaO.2Al₂O₃ or CaO.6Al₂O₃" should read -- CaO • 2Al₂O₃ or CaO • 6Al₂O₃ --.

Col. 12, line 11, "AlO₂" should read --AlO₂--.

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DATED : June 29, 2004

INVENTOR(S) : Paul W. Brown

It is certified that errors appear in the above-identified patent and that said Letters Patent should be corrected as shown below:

Col. 12, line 19, " $(5\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O})$ " should read $-(5\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O})-$.

Col. 12, line 22, " $\text{CaO} \cdot \text{Al}_2\text{O}_3$ " should read $-\text{CaO} \cdot \text{Al}_2\text{O}_3-$.

Col. 12, line 23, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3$ " should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3-$.

Col. 12, line 29, " $\text{CaO} \cdot \text{Al}_2\text{O}_3 + 3\text{Ca}(\text{OH})_2 + n\text{H}_2\text{O} \rightarrow 3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{OH})_2 \cdot n\text{H}_2\text{O}$ " should read
 $-\text{CaO} \cdot \text{Al}_2\text{O}_3 + 3\text{Ca}(\text{OH})_2 + n\text{H}_2\text{O} \rightarrow 3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{OH})_2 \cdot n\text{H}_2\text{O}-$.

Col. 12, lines 32 and 33, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{Cl})_2 \cdot n\text{H}_2\text{O}$ " should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{Cl})_2 \cdot n\text{H}_2\text{O}-$.

Col. 12, line 59, " $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$ " should read $-3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}-$.

Col. 13, line 48, "fiber" should read --further--.

Col. 14, lines 8, " $3\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}$ " should read
 $-3\text{Me}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}-$.

Col. 14, line 9, " $3\text{ME}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}$ " should read
 $-3\text{ME}(\text{II})\text{O} \cdot \text{R}_2\text{O}_3 \cdot \text{Me}(\text{II})(\text{anion})_2 \cdot n\text{H}_2\text{O}-$.

Col. 14, line 57, " $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$; and $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}$ " should read
 $-3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_2)_2 \cdot n\text{H}_2\text{O}$; and $3\text{CaO} \cdot \text{Fe}_2\text{O}_3 \cdot \text{Ca}(\text{NO}_3)_2 \cdot n\text{H}_2\text{O}-$.

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